

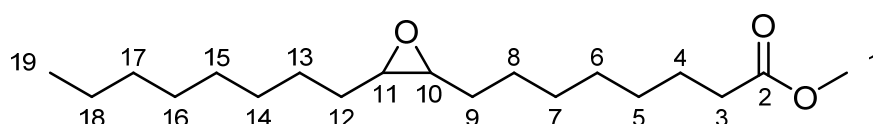
Electronic Supporting Information

Substrate dependent synergetic and antagonistic interaction of ammonium halide and polyoxometalate catalysts in the synthesis of cyclic carbonates from oleo epoxides and CO₂

Jens Langanke, Lasse Greiner and Walter Leitner

Analytic data:

Methyl 8-(3-octyloxiran-2-yl)octanoate **2**:

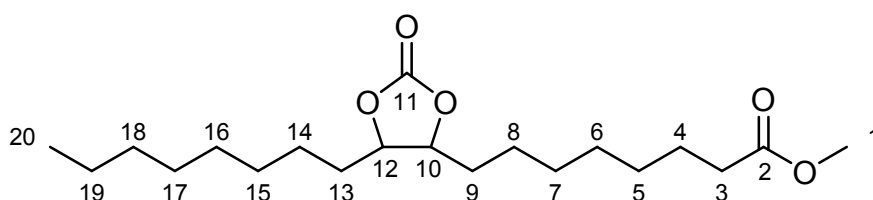


¹H-NMR (400 MHz, CDCl₃): δ = 0.81 (t, J =6.4 Hz, 3H, H19), 1.12-1.48 (m, 24H, H5-H9; H12-H18), 1.49-1.62 (m, 2H, H4), 2.23 (t, J =7.5 Hz, 2H, H3), 2.83 (brs, 2H, H10-H11), 3.59 (s, 3H, H1) ppm.

¹³C-NMR (100 MHz, CDCl₃): δ = 14.1 (C19, CH₃), 22.7 (C18, CH₂), 24.9 (C4, CH₂), 26.6 (CH₂), 26.6 (CH₂), 27.8 (CH₂), 27.8 (CH₂), 29.0 (CH₂), 29.2 (CH₂), 29.2 (CH₂), 29.3 (CH₂), 29.5 (CH₂), 29.6 (CH₂), 31.9 (C17, CH₂), 34.0 (C3, CH₂), 51.4 (C1, CH₃), 57.2 (C10, CH), 57.2 (C11, CH), 174.2 (C2, C) ppm.

GC (150 °C, 6 °C/min, 280 °C): rt = 13.5 min.

Methyl 8-(5-octyl-2-oxo-1,3-dioxolan-4-yl)octanoate **3**:



¹H-NMR (400 MHz, CDCl₃): δ = 0.88 (t, J =7.1 Hz, 3H, H20), 1.17-1.42 (m, 18H, H8-H9; H13-H19), 1.49-1.75 (m, 8H, H4-H7), 2.31 (t, J =7.5 Hz, 2H, H3), 3.67 (s, 3H, H1), 4.20-4.25 (m, 2H, *trans* H10, *trans* H12), 4.59-4.65 (m, 2H, *cis* H10, *cis* H12) ppm.

¹³C-NMR (100 MHz, CDCl₃): δ = 14.0 (C20, CH₃), 22.6 (C19, CH₂), 24.8 (C4, CH₂), 25.5 (C8, CH₂), 25.6 (C14, CH₂), 28.9 (CH₂), 28.9 (CH₂), 28.9 (CH₂), 29.0 (CH₂), 29.0 (CH₂), 29.1 (CH₂), 29.2 (CH₂), 29.3 (CH₂), 31.8 (CH₂), 34.0 (CH₂), 51.4 (C1, CH₃), 79.9 (C10, CH), 80.0 (C12, CH), 154.7 (C11, C), 174.2 (C2, C) ppm.

MS (EI): 356 (M⁺), 325, 239, 167, 153, 96, 84, 81, 57 m/z.

GC (150 °C, 6 °C/min, 280 °C): rt = 19.6 (*trans*-**3**), 20.2 (*cis*-**3**) min.

Identifying *cis/trans*-isomers of compound **3**:

The coupling partners of the protons H10 and H12 were identified using DGF-COSY spectra in C_6D_6 of both isolated *cis*- and *trans*-isomers of compound **3**. Selective decoupling of these protons in combination with molecular modelling of the two isomers using the software programs Spartan04 and Gaussian03 lead to the assignment of the signal at 3.60 ppm ($J = 5.4$ Hz) to *trans*-**3** and at 3.79 ppm ($J = 8.3$ Hz) to *cis*-**3** (spectra see below). The NMR-spectra were recorded in benzene- d_6 as solvent resulting in a shift of these signals in comparison to the spectra for compound **3** in $CDCl_3$.

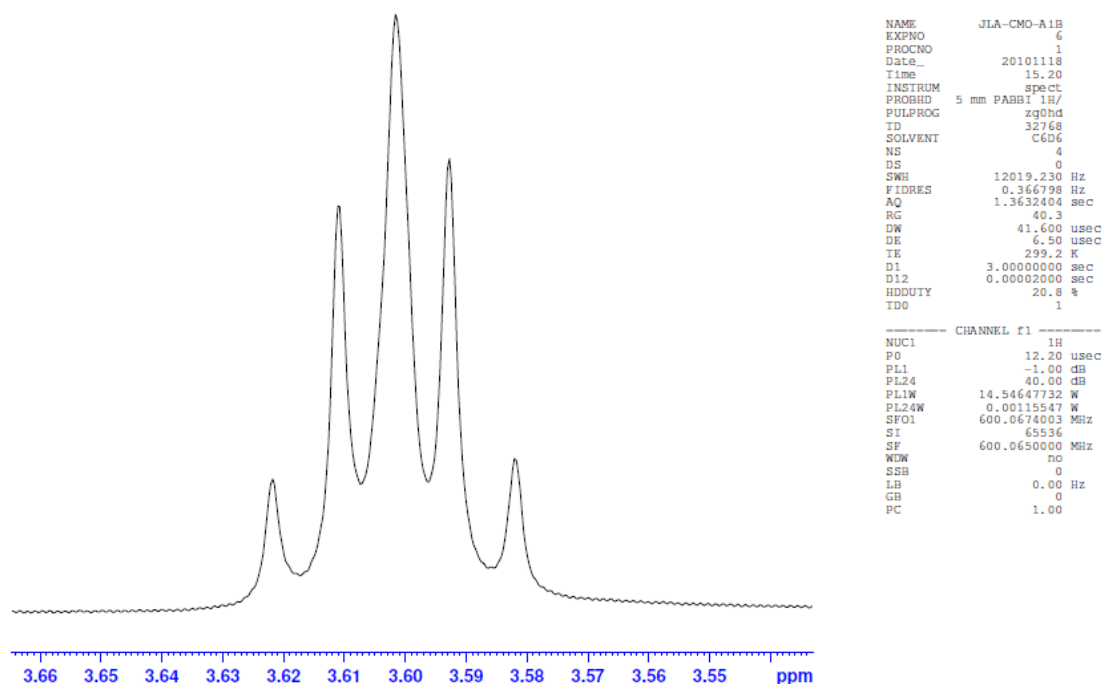
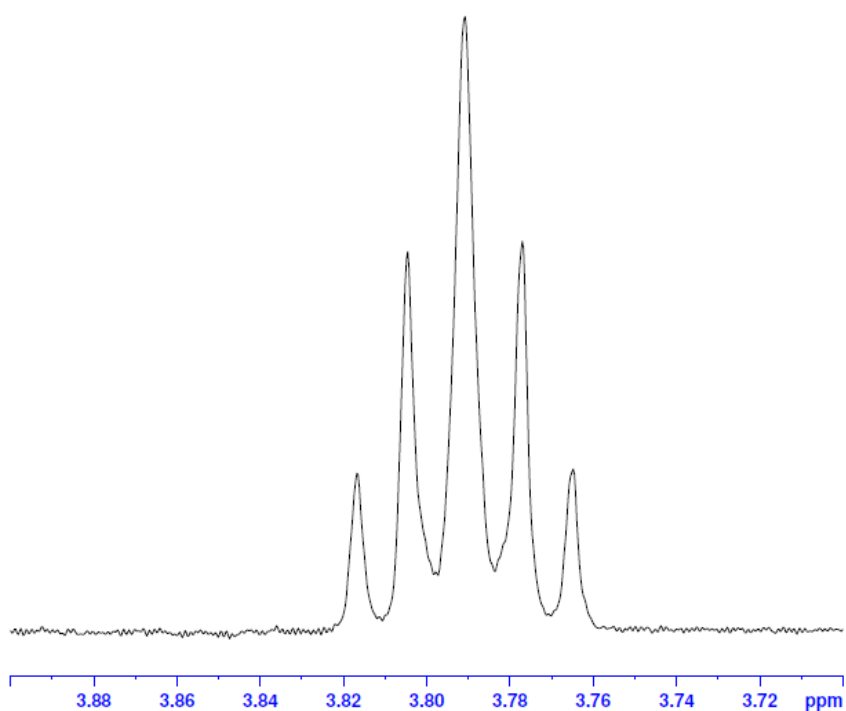


Figure ESI.3: Decoupled 1H -NMR spectrum of *trans*-**3** (section)

^1H @ QNP



```
NAME          JLA-CMOB1B
EXPNO         6
PROCNO       1
Date_        20101014
Time         13.31
INSTRUM      spect
PROBHD       5 mm FABBI 1H/
PULPROG      zgpg30
TD           32768
SOLVENT      CDCl3
NS           4
DS           0
SWH          12019.230 Hz
FIDRES       0.366798 Hz
AQ           1.3632404 sec
RG           40.3
DW           41.600 usec
DE           6.50 usec
TE           299.5 K
D1           3.00000000 sec
D12          0.00002000 sec
HDDUIY       20.8 %
TDO          1
```

```
----- CHANNEL f1 -----
NUC1          1H
PO           12.20 usec
PL1          -1.00 dB
PL24         40.00 dB
PL1W         14.54647732 W
PL24W        0.00115547 W
SFO1         600.0674003 MHz
SI           65536
SF           600.0650000 MHz
WDW          no
SSB          0
LB           0.00 Hz
GB           0
PC           1.00
```

Figure ESI.4: Decoupled ^1H -NMR spectrum of *cis*-**3** (section)

$((n\text{-C}_7\text{H}_{15})_4\text{N})_5[\text{CrSiW}_{11}\text{O}_{39}]$ **THA-Cr-Si-POM**:

MS (FAB-SIMS (+c/-c), Cs^+ , DTE/DTT matrix):

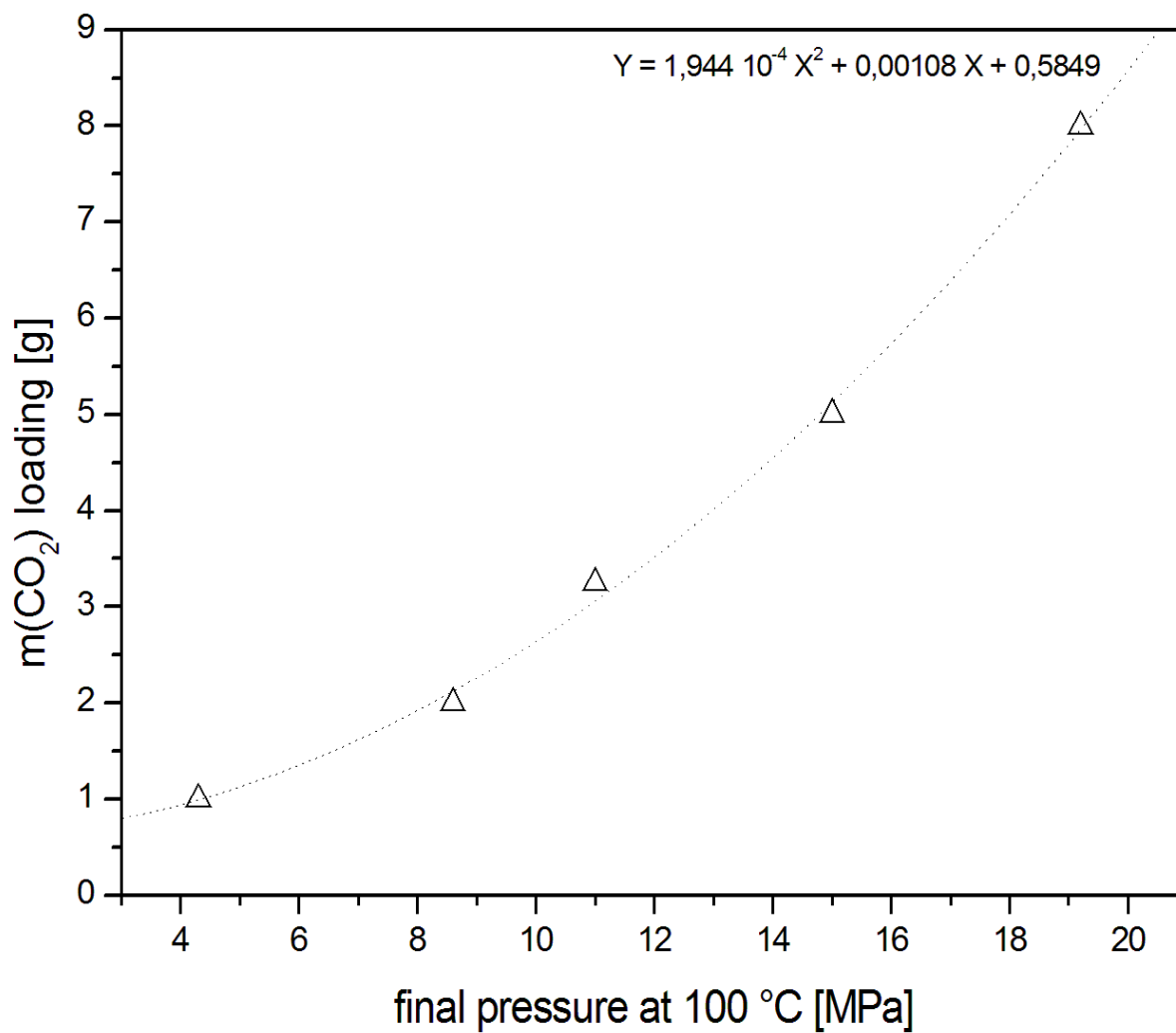
2727 (M⁻), 2695, 2463, 2216, 1984, 1752, 1518, 696,
481, 462;

411 (M⁺), 407. 311, 226, 142 m/z.

^{29}Si -NMR (120 MHz, Toluene- d_8): $\delta = -21.1$ ppm.

IR: $\nu_{\text{max}} = 2923$ (m), 2855 (w), 146 (w), 1378 (w), 952 (m), 907 (s), 780 (s), 523 (m) cm^{-1} .

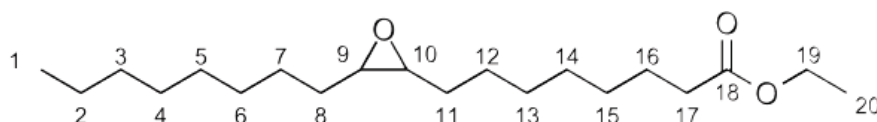
Calibration curve for charging the window-equipped stainless steel high-pressure reactor
(V = 11 ml) at room temperature with CO₂:



Synthesis procedures and analytic data:

1) Preparation of ethyl 8-(3-octyloxiran-2-yl)octanoate (**4**)

A 35 mL microwave vial was charged with 1 eq. ethyl (Z)-octadec-9-enoate (2.63 g, 8.47 mmol), and 1.2 eq. meta-chloroperoxybenzoic acid¹ (2.27 g, 77 %, 10.14 mmol) dissolved in 25 mL dichloromethan (DCM) was added slowly under ice cooling. The sealed vial was vigorously stirred and kept at 50 °C for 5 min by radiation with up to 10W power. The resulting reaction mixture was poured into 100 mL aq. NaHSO₃ solution (40 %) and stirred for 1 h. Afterwards, 115 mL aq. Na₂CO₃ solution (1 M) was added and it was again stirred for 1 h. A sample was withdrawn and tested for the absence of peroxides. The organic layer was collected, washed several times with water, pre-dried over anhydrous Na₂SO₄, and evaporated. The residue obtained was dissolved in 60 mL chloroform and again washed with water for several times. The organic layer was decanted and dried over anhydrous Na₂SO₄. After removing chloroform under reduced pressure, the pure product ethyl 8-(3-octyloxiran-2-yl)octanoate **4** was obtained as colourless oil (1.62 g, 4.97 mmol, 59 %).



Ethyl 8-(3-octyloxiran-2-yl)octanoate (**4**)

¹H-NMR (400 MHz, CDCl₃): δ = 0.88 (t, J=7.0 Hz, 3H, H1), 1.2-1.7 (m, 29 H, H2-H8, H11-H16, H20), 2.29 (t, J=7.5 Hz, 2 H, H17), 2.84-3.00 (m, 2 H, H9-H10), 4.12 (q, J=7.1 Hz, 2 H, H19) ppm.²

¹³C-NMR (100 MHz, CDCl₃): δ = 14.1 (C1, CH₃), 14.2 (C20, CH₃), 22.6 (C2, CH₂), 24.9 (C3, CH₂), 26.5-26.6 (C7, C12, 2 CH₂), 27.8 (C8, C11, 2 CH₂), 28.9-29.8 (C4-C6, C13-C15, 6 CH₂), 31.8 (C16, CH₂), 34.3 (C17, CH₃), 57.1-57.2 (C9, C10, 2 CH), 60.1 (C19, CH₂), 173.7 (C18, C) ppm.

IR: ν_{\max} = 2925 (m, C-H_{stretch}), 2855 (m), 1736 (s, C=O), 1465 (w), 1179 (m), 756 (s) cm⁻¹.

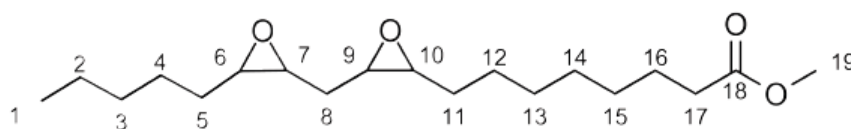
MS (EI, 70 eV): 326 (M⁺), 213, 157, 155, 153, 109, 101, 97, 95 m/z.

¹ mCPBA

² Y. Itoi, M. Inoue, S. Enomoto, *Bull. Chem. Soc. Jpn.*, 1986, **59**, 3941-3943.

2) Preparation of methyl 8-(3-((3-pentylloxiran-2-yl)methyl)oxiran-2-yl)octanoate (**6**)

A 35mL microwave vial was charged with 1 eq. (9Z,12Z)-methyl octadeca-9,12-dienoate (2.56 g, 8.68mmol), and 2.3 eq. mCPBA (4.54 g, 77 %, 20.24 mmol) dissolved in 30mL DCM was added slowly under ice cooling. The sealed vial was vigorously stirred and kept at 50 °C for 5 min by radiation with up to 10W power. The resulting reaction mixture was poured into 50mL aq. NaHSO₃ solution (40 %) and stirred overnight. The peroxide free DCM layer was decanted and stirred with 75 mL aq. Na₂CO₃ solution (1 M) for 0.5 h. After separating both phases, the aqueous one was extracted with DCM. The collected DCM phase was evaporated. The resulting residue was dissolved in 30mL chloroform and again treated with aq. Na₂CO₃ solution (50 mL, 1 M). After phase separation and extraction with either chloroform or aq. Na₂CO₃ solution, the chloroform phases were collected and dried over anhydrous Na₂SO₄. Finally, the chloroform was evaporated under reduced pressure and the pure product methyl 8-(3-((3-pentylloxiran-2-yl)methyl)oxiran-2-yl)octanoate (**6**) was obtained as colourless oil (2.09 g, 6.39 mmol, 74 %).



Methyl 8-(3-((3-pentylloxiran-2-yl)methyl)oxiran-2-yl)octanoate (**6**)

¹H-NMR (400 MHz, CDCl₃): δ = 0.90 (t, J=6.9 Hz, 3 H, H1), 1.20-1.84 (m, 22 H, H2-H5, H8, H11-H16), 2.30 (t, J=7.5Hz, 2 -H, H17), 2.92-3.15 (m, 4 H, H6, H7, H9, H10), 3.66 (s, 3 H, H19) ppm.³

¹³C-NMR (100 MHz, CDCl₃): δ = 14.0 (C1, CH₃), 22.6 (C2, CH₂), 24.9-27.9 (C3, C4, C5, C8, C11, C12, 6 CH₂), 29.0-29.3 (C13-C15, 3 CH₂), 31.7 (C16, CH₂), 34.0 (C17, CH₂), 51.4 (C19, CH₃), 54.2-54.3 (C7, C9, 2 CH), 56.7-57.0 (C6, C10, 2 CH), 174.2 (C18, C) ppm.³

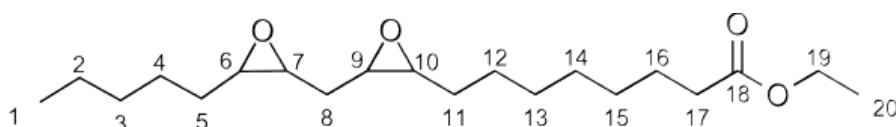
IR: ν_{max} = 2930 (m, C-H_{stretch}), 1737 (s, C=O), 1436 (w), 1170 (m), 753 (s) cm⁻¹.

MS (EI, 70 eV): 326 (M+), 155, 123, 109, 98, 97, 96, 95, 93 m/z.

³ K. M. Doll, S. Z. Erhan, *J. Agric. Food Chem.*, 2005, **53**, 9608-9614.

3) Preparation of ethyl 8-(3-((3-pentylloxiran-2-yl)methyl)oxiran-2-yl)octanoate (**8**)

A 35 mL microwave vial was charged with 1 eq. of (9Z,12Z)-ethyl octadeca-9,12-dienoate (1.42 g, 4.60 mmol), and 3.3 eq. mCPBA (3.36 g, 77 %, 15.00 mmol) dissolved in 25 mL chloroform was added slowly under ice cooling. The sealed vial was vigorously stirred and kept at 50 °C for 5 min by radiation with up to 10W power. The same procedure was repeated with a second identical batch. Both batches were combined and stirred for 1 h with 100 mL aq. NaHSO₃ solution (40 %). The peroxide-free chloroform layer was decanted and stirred for 1 h with 100mL aq. Na₂CO₃ solution (1 M). After separating both phases, the aqueous one was extracted with 100 mL chloroform and the collected chloroform phase was extracted once with 50 mL aq. Na₂CO₃ solution (1 M) and twice with each 100 mL water. The combined chloroform phase was pre-dried over anhydrous Na₂SO₄ and evaporated. The resulting residue was dissolved in 70 mL chloroform, washed six times with each 50 mL water, separated, and dried over anhydrous Na₂SO₄. Finally, the chloroform was evaporated under reduced pressure and the pure product ethyl 8-(3-((3-pentylloxiran-2-yl)methyl)oxiran-2-yl)octanoate was obtained as turbid oil (2.06 g, 6.06 mmol, 66 %).



Ethyl 8-(3-((3-pentylloxiran-2-yl)methyl)oxiran-2-yl)octanoate (**8**)

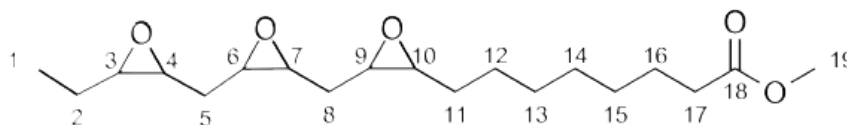
¹H-NMR (400 MHz, CDCl₃): δ = 0.83 (t, J=6.6 Hz, 3 H, H1), 1.15_1.80 (m, 25 H, H2-H5, H8, H11-H16, H20), 2.22 (t, J=7.6 Hz, 2 H, H17), 2.86-3.08 (m, 4 H, H6, H7, H9, H10), 4.05 (q, J=7.1 Hz, 2 H, H19) ppm.

¹³C-NMR (100 MHz, CDCl₃): δ = 14.0 (C1, CH₃), 14.2 (C20, CH₃), 22.6 (C2, CH₂), 24.9 (C3, CH₂), 26.1 (C8, CH₂), 26.5 (C4, CH₂), 27.1 (C12, CH₂), 27.8-27.9 (C5, C11, 2 CH₂), 29.0-29.3 (C13-C15, 3 CH₂), 31.7 (C16, CH₂), 34.3 (C17, CH₂), 54.2-54.3 (C7, C9, 2 CH), 56.7-57.0 (C6, C10, 2 CH), 60.2 (C19, CH₂), 173.8 (C18, C) ppm.

MS (EI, 70 eV): 341 (M⁺), 217, 155, 109, 98, 97, 96, 95 m/z.

4) Preparation of methyl 8-(3-((3-((3-ethyloxiran-2-yl)methyl)oxiran-2-yl)methyl)oxiran-2-yl)octanoate (**10**)

A 35mL microwave vial was charged with 1 eq. of (9Z,12Z,15Z)-methyl octadeca-9,12,15-trienoate **24** (1.14 g, 3.91 mmol), and 3.6 eq. mCPBA (3.13 g, 77 %, 13.98 mmol) dissolved in 25 mL chloroform was added slowly under ice cooling. The sealed vial was vigorously stirred and kept at 50 °C for 5 min by radiation with up to 10W power. The same procedure was repeated with a second identical batch. Both batches were combined and stirred for 1 h with 100 mL aq. NaHSO₃ solution (40 %). The peroxide-free chloroform layer was decanted and stirred with 100mL aq. Na₂CO₃ solution (1 M) for 1 h. After separating both phases, the aqueous one was extracted with 100mL chloroform and the collected chloroform phase was extracted twice with each 100 mL water. The combined chloroform phase was dried over anhydrous Na₂SO₄. Finally, the chloroform was evaporated under reduced pressure and the pure product methyl 8-(3-((3-((3-ethyloxiran-2-yl)methyl)oxiran-2-yl)methyl)oxiran-2-yl)octanoate (**10**) was obtained as yellowish oil (2.38 g, 6.98 mmol, 89 %).



Methyl 8-(3-((3-((3-ethyloxiran-2-yl)methyl)oxiran-2-yl)methyl)oxiran-2-yl)octanoate (**10**)

¹H-NMR (400 MHz, CDCl₃): δ =0.99 (t, J=7.4 Hz, 3 H, H1), 1.2-1.8 (m, 18 H, H2, H5, H8, H11-H16), 2.24 (t, J=7.5 Hz, 2 H, H17), 2.85-3.16 (m, 6 H, H3, H4, H6, H7, H9, H10), 3.60 (s, 3 H, H19) ppm.

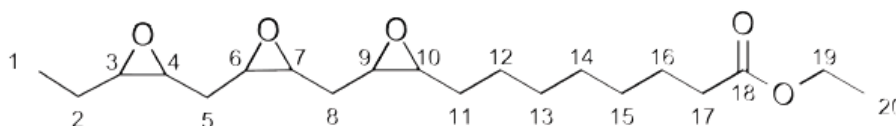
¹³C-NMR (100 MHz, CDCl₃): δ =10.5 (C1, CH₃), 21.2 (C2, CH₂), 24.9 (C12, CH₂), 26.1-27.9 (C3, C5, C8, 3 CH₂), 27.8 (C11, CH₂), 29.0-29.3 (C13-C15, 3 CH₂), 34.0 (C17, CH₂), 51.5 (C19, CH₃), 53.8-58.3 (C4, C6, C7, C9, C10, C16, 6 CH), 174.2 (C18, C) ppm.

IR: ν_{max} =2928 (m, C-H_{stretch}), 1736 (s, C=O), 1436 (w), 1169 (m), 821 (m) cm⁻¹.

MS (EI, 70 eV): 340 (M⁺), 215, 157, 155, 109, 97, 93, 91 m/z.

5) Preparation of ethyl 8-(3-((3-((3-ethyloxiran-2-yl)methyl)oxiran-2-yl)methyl)oxiran-2-yl)octanoate (**12**)

A 35mL microwave vial was charged with 1 eq. (9Z,12Z,15Z)-ethyl octadeca-9,12,15-trienoate **26** (1.25 g, 4.07 mmol), and 4.5 eq. mCPBA (4.07 g, 77 %, 18.17mmol) dissolved in 25 mL chloroform was added slowly under ice cooling. The sealed vial was vigorously stirred and kept at 50 °C for 5 min by radiation with up to 10W power. The same procedure was repeated with a second identical batch. Both batches were combined and stirred for 2 h with 150 mL aq. NaHSO₃ solution (40 %). The peroxide-free chloroform layer was decanted and stirred for 1 h with 100mL aq. Na₂CO₃ solution (1 M). After separating both phases, the aqueous one was extracted with 100 mL chloroform and the collected chloroform phase was extracted with 75mL aq. Na₂CO₃ solution (1 M) and six times with each 100mL water. The combined chloroform phase was dried over anhydrous Na₂SO₄. Finally, the chloroform was evaporated under reduced pressure and the pure product ethyl 8-(3-((3-((3-ethyloxiran-2-yl)methyl)oxiran-2-yl)methyl)oxiran-2-yl)octanoate (**12**) was obtained as turbid oil (1.27 g, 3.58 mmol, 44 %).



Ethyl 8-(3-((3-((3-ethyloxiran-2-yl)methyl)oxiran-2-yl)methyl)oxiran-2-yl)octanoate (**12**)

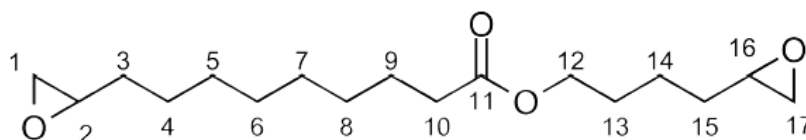
¹H-NMR (400 MHz, CDCl₃): δ = 0.99 (t, J=7.5 Hz, 3 H, H1), 1.13-1.81 (m, 21 H, H2, H5, H8, H11-H16, H20), 2.22 (t, J=7.5 Hz, 2 H, H17), 2.81-3.15 (m, 6 H, H3, H4, H6, H7, H9, H10), 4.06 (q, J = 3.6 Hz, 2 H, H19) ppm.

¹³C-NMR (100 MHz, CDCl₃): δ = 10.5 (C1, CH₃), 14.3 (C20, CH₃), 21.3 (C2, CH₂), 24.9 (C3, CH₂), 26.1-28.0 (C5, C8, C12, 3 CH₂), 28.9-29.3 (C13-C15, 3 CH₂), 34.3 (C17, CH₂), 54.2 (C16, CH), 56.6-57.0 (C4, C6, C7, C9, 4 CH), 58.2 (C10, CH), 60.2 (C19, CH₂), 173.8 (C18, C) ppm.

MS (EI, 70 eV): 354 (M⁺), 213, 157, 110, 108, 107, 98, 96, 95 m/z.

6) Preparation of 4-(oxiran-2-yl)butyl 9-(oxiran-2-yl)nonanoate (**14**)

A 35mL microwave vial was charged with 1 eq. hex-5-enyl undec-10-enoate (1.33 g, 4.97 mmol), and 2.8 eq. mCPBA (3.11 g, 77 %, 13.88 mmol) dissolved in 25 mL chloroform was added slowly under ice cooling. The sealed vial was vigorously stirred and kept at 50 °C for 5 min by radiation with up to 10W power. The same procedure was repeated with a second identical batch. Both batches were combined and stirred for 1 h with 100mL aq. NaHSO₃ solution (40 %). The peroxide-free chloroform layer was decanted and stirred for 40 min with 100 mL aq. Na₂CO₃ solution (1 M). After separating both phases, the aqueous one was extracted with 100 mL chloroform and the collected chloroform phase was extracted with 50 mL aq. Na₂CO₃ solution (1 M) and twice with each 100 mL water. The combined chloroform phase was dried over anhydrous Na₂SO₄. Finally, the chloroform was evaporated under reduced pressure and the pure product 4-(oxiran-2-yl)butyl 9-(oxiran-2-yl)nonanoate (**14**) was obtained as viscous colourless liquid (2.44 g, 8.18 mmol, 82 %).



4-(oxiran-2-yl)butyl 9-(oxiran-2-yl)nonanoate (**14**)

¹H-NMR (400 MHz, CDCl₃): δ = 1.16-1.73 (m, 20 H, H3-H9, H13-H15), 2.22 (t, J=7.5 Hz, 2 H, H10), 2.37-2.42 (m, 2 H, H1, H17), 2.65-2.71 (m, 2 H, H2, H16), 2.80-2.88 (m, 2 H, H1, H17), 4.00 (t, J = 6.5 Hz, 2 H, H12) ppm.

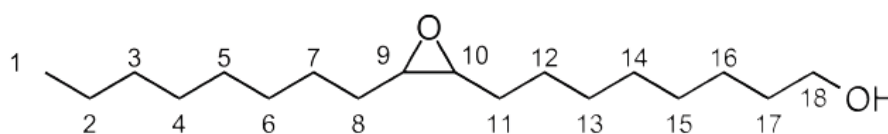
¹³C-NMR (100 MHz, CDCl₃): δ = 22.5 (C14, CH₂), 24.9 (C9, CH₂), 25.9 (C4, CH₂), 28.4 (C13, CH₂), 29.1-29.3 (C5-C8, CH₂) 32.1 (C15, CH₂), 32.5 (C3, CH₂), 34.3 (C10, CH₂), 47.0-47.1 (C1, C17, CH₂), 52.1-52.4 (C2, C16, CH), 64.0 (C12, CH₂), 173.9 (C11, C) ppm.

IR: ν_{max} = 2926 (m, C-H_{stretch}), 2855 (m), 1732 (s, C=O), 1459 (w), 1170 (m), 914 (w), 835 (m) cm⁻¹.

MS (EI, 70 eV): 298 (M⁺), 139, 123, 121, 99, 98, 97, 95 m/z.

7) Preparation of 8-(3-octyloxiran-2-yl)octane-1-ol (**18**)

A 35 mL microwave vial was charged with 1 eq. (Z)-octadec-9-en-1-ol (2.65 g, 9.96 mmol), and 1.6 eq. mCPBA (3.63 g, 77 %, 16.23 mmol) dissolved in 25 mL chloroform was added slowly under ice cooling. The sealed vial was vigorously stirred and kept at 50 °C for 5 min by radiation with up to 10W power. The reaction mixture was poured into 50mL aq. NaHSO₃ solution (40 %) and stirred for 20 min. Afterwards, 50mL aq. Na₂CO₃ solution (1 M) was added and stirred for another 40 min. The peroxide-free chloroform layer was decanted and washed five times with each 200 mL water. The whole procedure was repeated with a second identical batch. The resulting chloroform phases were combined and dried over anhydrous Na₂SO₄. Finally, the chloroform was evaporated under reduced pressure and the pure product 8-(3-octyloxiran-2-yl)octane-1-ol (**18**) was obtained as beige waxy solid (2.89 g, 10.16 mmol, 51 %).



8-(3-octyloxiran-2-yl)octane-1-ol (**18**)

¹H-NMR (400 MHz, CDCl₃): δ =0.88 (t, J=6.5 Hz, 3 H, H1), 1.2-1.66 (m, 28 H, H2-H8, H11-H17), 2.85-2.94 (m, 2 H, H9, H10), 3.63 (t, J=6.5 Hz, 2 H, H18) ppm.

¹³C-NMR (100 MHz, CDCl₃): δ =14.1 (C1, CH₃), 22.7 (C2, CH₂), 25.7 (C3, CH₂), 26.6-29.7 (C4-C8, C11-C15, 8 CH₂), 31.9 (C16, CH₂), 32.8 (C17, CH₂), 57.3 (C9, C10, 2 CH), 63.0 (C18, CH₂) ppm.

IR: ν_{max} =3359 (br, O-H_{stretch}), 2917 (s, C-H_{stretch}), 1460 (m, O-H_{deformation}), 1074 (s, C-OH), 979 (s), 845 (s), 632 (m) cm⁻¹.

MS (EI, 70 eV): 284 (M⁺), 171, 155, 125, 124, 123, 112, 111, 98, 97, 96, 95 m/z.