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A traveling wave decelerator for neutral polar molecules

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Recently, a decelerator for neutral polar molecules has been presented that operates on the basis of macroscopic, three-dimensional, traveling electrostatic traps [A. Osterwalder, S. A. Meek, G. Hammer, H. Haak, and G. Meijer, Phys. Rev. A 81, 051401 (2010)]. In the present paper, a complete description of this decelerator is given, with emphasis on the electronics and the mechanical design. Experimental results showing the transverse velocity distributions of guided molecules are shown and compared to trajectory simulations. An assessment of non-adiabatic losses is made by comparing the deceleration signals from $^{13}$CO with those from $^{12}$CO and with simulated signals.


I. INTRODUCTION

Interest in the production of cold molecules stems from a variety of research areas, most notably collision dynamics, the study of long-range interactions, quantum computation, and high-resolution spectroscopy (for a recent overview, see the individual chapters of Ref. 1).

Two principal approaches can be distinguished in the production of cold molecules: either the molecules are assembled from ultracold atoms using electric fields, magnetic fields, or lasers, or one starts from warm or fast samples and brings them to low average translational temperatures by controlling the velocity.1–2 Variants of the latter approach include using collisions with a cold buffer gas3 or between molecules in crossed-beams,4 by mechanical means such as rotating nozzles,5,6 or through the controlled interaction with electric,7–9 magnetic,10–12 or optical fields.13

One simple way to use electric fields to passively control a molecular velocity distribution is by filtering out fast molecules using the competition between an electrostatic force and a centrifugal force in a bent guide.14–16 This does not change the velocity of the individual molecules but merely extracts the slow ones from a thermal distribution. Active control of the forward velocity of molecules with electric fields was first demonstrated in 1999,7,8 rapidly followed by the first trapping of neutral molecules.17,18 Further developments of this technology have enabled the deceleration of large molecules in high-field-seeking states,19 the trapping of molecules in static and ac-fields,18,20 and the storage of molecules in a synchrotron.21,22

For most of the studies mentioned above, considerable density both in velocity space and in position space are of great importance. A significant development in Stark deceleration was, thus, the recognition that such devices should be operated in a higher-order mode in order to suppress the coupling of longitudinal and transverse oscillations and the losses following from them.23,24 Recently, a new approach has been presented to completely avoid this problem, namely, the use of traveling three-dimensional traps to capture and decelerate molecules.25 This approach builds on ideas implemented in a miniaturized Stark decelerator on a chip,26–28 and a similar principle has also been applied to the deceleration of Rydberg states.29 In contrast to a conventional Stark decelerator, where effective moving traps are obtained by periodic switching of the electric fields between static configurations, these devices employ continuous modulation of the voltages to produce real, continuously moving three-dimensional traps. In the chip decelerator, this modulation is chosen to produce a chain of traps above a periodic structure of electrodes, which in this case are stripes on a dielectric substrate. The modulation frequency determines, for a particular electrode geometry, the velocity of the traps and can be chirped to produce acceleration or deceleration. In the present decelerator, the stripes of the chip decelerator are replaced with three-dimensional ring electrodes.25 The decelerator is scaled up by a factor of 100 in comparison with the chip decelerator, which results in dimensions comparable to conventional Stark decelerators. Based on trajectory calculations, the six-dimensional phase space acceptance for guiding of OH molecules at a constant velocity is determined to be a factor of three larger than in the state-of-the-art switched Stark decelerator,24 and during deceleration, the same relative phase space acceptance is maintained for an acceleration which is a factor of three larger. Furthermore, the molecules are loaded into the traps at high velocity, and the loading of stationary traps is simply bringing the traveling traps to standstill.

The present paper gives a detailed account of the design and construction of this new decelerator and shows experimental data together with the corresponding trajectory simulations to explain the operation principle of the device and the dynamics of individual molecules inside the decelerator. The paper is structured as follows: Sec. II describes the general principle of operation (Sec. II A), the mechanical design (Sec. II B), and the required electronics (Sec. II C).
Section II B includes the particular approach chosen to assemble the structure. Section III presents experimental results that characterize the apparatus and theoretical calculations to support the interpretation of these measurements. The paper is summarized in Sec. IV where also an outlook is given, along with future applications of this design.

II. EXPERIMENTAL

Sections II A–II C give a general overview on the operation principle of the decelerator and describe in detail how the device is designed and constructed and how the required electronics are built.

A. General considerations

Applying a spatial sine-modulated electric potential to a periodic array of ring electrodes produces an electric field distribution on the inside of the array with two on-axis minima in the field magnitude per period (see Figure 1). Due to the cylindrical symmetry, these minima represent three-dimensional traps for polar molecules.

The required potential on the individual electrodes can be expressed as

\[ V_n(t) = V_0 \sin \left( -\phi_0(t) + \frac{2\pi n}{N} \right), \]

with the waveform amplitude \( V_0 \) and the time-dependent phase offset \( \phi_0(t) \). The electrode-dependent shift is determined through the periodicity \( N \) of the specific array. A linear increase of \( \phi_0(t) \) with time continuously shifts the traps along the cylinder axis. The time-dependence of the phase offset can be expressed as

\[ \phi_0(t) = 2\pi \int_0^t v(\tau) d\tau, \]

where the frequency \( v(\tau) \) is the modulation frequency of the voltage on each individual electrode. Since one oscillation of the waveform moves the trap over one period, the velocity of the traps is given by \( L v(\tau) \), where \( L \) is the period length. The time-dependence of the frequency leads to acceleration or deceleration of the traps. After calculating the acceptance for many different geometrical parameters, a suitable combination of these was found with 4 mm inner electrode diameter, 0.6 mm electrode thickness, 12 mm period length, and periodicity 8. Lower periodicity means fewer points to sample the spatial sine, and thus less smooth electric field distributions. Higher periodicity improves the electric fields but makes the setup technically more challenging. The field distribution produced by a larger number of electrodes per period alone is not necessarily better: in optimizing the number of electrodes, it was found that – while keeping all other parameters the same – a decelerator with 8 electrodes per period performs better than one with 12 and also one with 6, while a decelerator with 10 electrodes per period performs about the same. With constant electrode thickness, a higher periodicity leads to reduced distortions in the field distribution near the electrodes. When going from 8 electrodes to 6, these distortions do not decrease further, but the field strength in the region around the central axis decreases. These observations indicate that there is an optimum ratio between the thickness of the electrodes and the width of the gap between them.

In order to create sufficiently deep traps moving at the required velocity for this geometry, the electronics must produce waveforms with \( V_0 \approx 10 \text{kV} \), with a tunable frequency up to \( \approx 25 \text{kHz} \) (corresponding to a velocity of 300 m/s). The phase of the waveforms between neighboring electrodes is shifted by \( \pi/4 \).

Contour plots for single traps resulting from \( V_0 = 10 \text{kV} \) sine potentials used with this design are shown in Figures 1(b) and 1(d) for the cases where the trap is positioned in the plane of an electrode (corresponding to \( \phi_0 = 0 \)) and between two electrodes (\( \phi_0 = \pi/8 \)), respectively. Contours are spaced by 5 kV/cm, starting at 5 kV/cm. Thick lines show the contours for 10 kV/cm and 20 kV/cm, respectively. The gray circles
to the right of panels (b) and (d) show the positions of the electrodes. Numbers next to the circles show the voltages that are applied to each electrode in the two cases. Panels (a) and (c) show the radial dependence of the electric field magnitude at positions designated by the lines with corresponding line types in panels (b) and (d). Each line represents the cross section for different distances from the trap minimum. In both cases, the solid, dotted, dashed-dotted, long-dashed, and long-short-dashed lines represent 0 mm, 0.75 mm, 1.5 mm, 2.25 mm, and 3 mm from the trap minimum; 3 mm from the trap minimum is the position of the saddle point toward the next trap. The on-axis longitudinal potential is shown in panel (e) for the two cases. Here, the solid (dashed) line represents the case with φ₀ = 0 (φ₀ = π/8). Panel (e) shows that, longitudinally, the potential does not change when moving the trap from in-plane to between two electrode planes. This is also confirmed through the height of the saddle point in panels (a) and (c). Longitudinally, the traps are almost 40 kV/cm deep when sine waves with amplitudes of 10 kV are applied to the electrodes. For a linearly Stark-shifted state, the trap has the shape of | sin(x) |, which is nearly linear close to the minimum. While the shape of the longitudinal potential is very stable and independent of φ₀, the radial potential does undergo some oscillations. The trap is almost 50 kV/cm deep when located in the plane of an electrode, but ∼13 kV/cm when between two electrodes. For the a 3Π₁ (v = 0, J = 1) state of CO used here this leads to a transverse acceptance of ∼±22 m/s and ∼±11 m/s, respectively. Trajectory simulations on a virtual decelerator with 10 m length demonstrated that the effective transverse acceptance lies in between these values and is close to the one obtained when averaging the radial potential along the guide axis. Longitudinally, the acceptance for guiding at constant velocity is ∼±19 m/s.²⁵

**B. Mechanical design and construction**

The upper part of Figure 2 is a photograph of the top end of the decelerator. The central part shows the electrodes, visible as horizontal bars, and the electrode mounting rods, as described below. The lower part of the photograph shows the mounting structure with the main octagonal stainless-steel tube and bronze plates that hold the electrode mounts. An overview of the decelerator design, including its mount, is shown in Figure 2(a). Figures 2(b)–2(d) show, respectively, a top view of the decelerator, a single electrode mount, and an electrode with a cross section through the mount. The decelerator is positioned vertically inside a high-vacuum chamber.

Molecules are produced in a pulsed supersonic expansion and enter the decelerator chamber through a skimmer, visible at the bottom of Figure 2(a), that is placed as close as possible to the beginning of the decelerator. In the case of the CO experiments presented below, a UV laser crosses the expansion directly before the skimmer to excite the CO molecules to the upper Λ-doublet component of the metastable a 3Π₁ (v = 0, J = 1) state. This particular state has a phosphorescence lifetime of 2.63 ms.²⁰

The decelerator is formed from 320 ring electrodes that are mounted on 8 separate 6 mm-diameter stainless steel rods. It is worth remembering that the number of electrodes here can not be compared to the number of deceleration stages in a traditional Stark decelerator; the figure of merit is instead the electric field gradient that can be generated, and this determines the length of the decelerator, which in turn defines the number of electrodes. The 8 rods are each mounted on ceramic posts that are attached via adjustable bronze bars to the main octagonal mounting structure (see Figure 2(b)). The mechanical point of reference of the entire setup is the center of the base flange that holds the skimmer and supports the decelerator. This construction, and the vertical orientation of the decelerator, greatly simplifies the alignment by making it independent of the main vacuum chamber. A magnification of one of the rods is shown in panel 2(c). The individual electrodes are spaced by 12 mm which defines the period length. All 8 rods, hand-picked for straightness, are identical.
and are mounted with a longitudinal offset of 1.5 mm from their nearest neighbors, a value which is given by the required center-to-center distance of neighboring electrodes. The arrangement of successive electrode rods is counter-clockwise when seen from the skimmer, forming a left-handed helical pattern.

A single electrode is shown in panel 2(d). The active part of these electrodes is the top, circular part with an inner diameter of 4 mm. The bottom part is the handle via which the electrode is mounted to the main rod. Each electrode is formed from 0.6 mm diameter Ta wire in a three-step procedure:

1. The wire is stretched, polished, and cut to the approximate length required for the electrode;
2. Each piece is bent into a rectangular “U” shape by forcing it around the corners of a 13 mm (~π 4 mm, found by trial and error) hardened steel plate;
3. In order to obtain the inner circular shape, each electrode is bent around a 4 mm stainless steel rod.

Tantalum wire is used because of its high resilience to electrical discharges and because it can be bent into the desired shape quite easily. The stretching of the wire is necessary to ensure straight wires, but it also reduces the diameter to 0.57 mm. Step 2 is crucial because it ensures that the wires are bent at right angles at the point where the two ends of the circular bend meet. All attempts to bend the electrodes directly into this shape, without the pre-bending, resulted in electrodes with cross sections that resembled teardrops rather than circles. Considerable deviation from the overall cylindrical symmetry of the decelerator is obtained if the individual electrodes are not as close to circular as possible. The above procedure guarantees that all 320 electrodes are identical to within high accuracy, and that the electrodes are produced without being scratched, thus rendering them more resistant to discharges when high voltages are applied. The electrodes are then cleaned in an ultrasonic bath and mounted on the main electrode rods. Cleaning is an important step in order to remove traces of the polishing material from the surface of the electrodes, and this step significantly enhances the resilience of the electrodes towards electrical discharges.

Each of the mounting rods contains 40 holes for the electrodes, one of which is seen in Figure 2(d), and two tapped holes for the mounting. The ring electrodes are pushed into the holes and pre-aligned using a polyvinyl chloride (PVC) block that fixes the distance between the rod surface and the electrode bend. The Ta electrodes are then fixed to the steel rods by pushing one or two pieces of bent, 0.3 mm piano wire into the spaces left between the Ta wire and the walls of the hole. This ensures that the electrodes do not slip out but leaves the required flexibility for final alignment. At this stage, the electrodes are positioned to within ~0.5 mm, insufficient for the present purposes.

The exact positioning of the electrodes on the individual rods is complicated by the fact that their positions have to be measured without mechanical contact, both in order to avoid damage on the surface of the electrodes, and because any contact might bend the wires. A solution is obtained by pre-mounting the rod with the ceramic mounts and the bronze bar (in the same combination as in the final setup) on an aluminum block. This block is then positioned on a digital microscope that is built by fixing a CCD camera equipped with a macro-zoom lens to the head of a mill. The aluminum block, together with the pre-mounted electrode assembly, is accurately positioned on the moving table of the mill. In this way, the relative positions of individual electrodes on each rod can be measured and set with a limit given by the resolution of the CCD. By placing the mount with the electrodes horizontal or vertical, it is possible to align and position the electrodes relative to the main rod and relative to each other with an estimated accuracy of ±0.02 mm.

Finally, the eight rods are mounted on the main cylinder and aligned to each other. This last step is done by positioning one CCD camera above and one to the side of the decelerator, and a theodolite below it. Using custom-made alignment tools, each rod is positioned relative to its neighboring rods and to the main axis of the decelerator. The vertical alignment can only be measured at one particular point, but the very accurate positioning of the individual electrodes on the mounting rod ensures the same degree of alignment throughout the entire structure.

A quantitative assessment of the overall precision in the alignment is difficult. Table I summarizes the main estimated deviations from the ideal structure. These estimates base on a critical evaluation of each of the production steps, as well as on test measurements on selected electrodes and electrode positions after the alignment. The electrode diameter is defined through the bending around the 4 mm rod, and the deviation stated in the table has to be read as possible deviations from a perfectly round shape, as well as a deviation from the ideal diameter. Due to the use of optical tools in the alignment procedure, the most difficult, and thus also least precise, part is the accurate positioning of the main electrode mounting rods in the plane perpendicular to the guide axis, mainly because of the mutual visual obstruction of the electrodes.

An attempt to characterize the effects of these deviations has been made by simulating the resulting electric fields in three dimensions using finite element calculations. A horizontal displacement of a single electrode moves the trap off-center when it is located in the plane of that electrode and has little effect otherwise. Displacement of an entire rod, thus, leads to a periodic jumping of the traps, with a frequency given by the modulation frequency. Vertical displacement changes the electric field between two electrodes but has
III. RESULTS

The operation of the apparatus is demonstrated in Secs. III A–III D. In Sec. III A, the phosphorescence of the metastable CO molecules is used to image their motion along the decelerator axis and the formation of density gradients near the trap positions. In Sec. III B, the transverse distributions are investigated with application of dc voltages to the electrodes, thus, forming an electrostatic guide. Section III C describes the results of the imaging of molecules that were guided in modulated electric fields. Finally, the effect of potential non-adiabatic transitions in CO is studied by comparing arrival time distributions of $^{12}$CO and $^{13}$CO. These results complement the measurements presented previously. There, arrival time distributions for $^{13}$CO were presented for guiding with modulated potentials and for deceleration.

A. Phosphorescence images of molecules in the decelerator

The metastable state of CO used here has a lifetime of 2.63 ms and it decays to the electronic ground state by emission of a photon with wavelength between 206 nm and 240 nm. This phosphorescence can be imaged, using an image-intensified CCD camera, to follow the density of CO molecules as a function of time and position.

Figure 4 shows images taken from the side of the decelerator about 15 cm from its entrance (see Figure 2(a)). These images were taken through the gaps between the decelerator electrodes, and the shadows of the electrodes are visible as a periodic intensity modulation. The molecules were guided through the decelerator at a constant velocity of 300 m/s. Figure 4(a) shows a single 2 µs-long snapshot recorded at a fixed delay of 617 µs relative to the laser pulse that produces the metastable CO molecules. This delay was chosen such that the densest part of the distribution would be located exactly between two electrodes. The horizontal axis corresponds to the molecular beam axis, with the source on the left and the detector on the right. The graph above the image shows the same data integrated over the transverse coordinate. The intense region at the center of the image is produced by the single trap that is filled by the molecular beam; the false color image of the electric field magnitude at the bottom of the panel shows the position of the traps (darker areas) at the time of the snapshot. On either side of the main peak are molecules that jumped over the barrier to the previous or next trap. Because of their energy, they continue to pass over all saddle points along the decelerator. Since they slow down on the saddle points, there is a slight increase in the density which is visible six electrode spacings away on either side of the main peak. The structure becomes more pronounced by the time the molecules have passed through the entire decelerator, and they are very clear in the arrival time distributions shown, e.g., in Figure 7. The detection efficiency of the imaging is relatively low. The 5 cm-diameter lens 35 cm from the center of the decelerator only collects 0.13% of the light emitted by the phosphorescing molecules, and the imaging system itself is only about 10% efficient. Additionally, during a 2 µs exposure, only about 0.07% of the molecules will...
The resulting, overall detection efficiency is roughly $10^{-7}$. From this and the total number of photons observed in the images, we can infer that about $10^7$ molecules are guided in the main trap at a density of $10^{10}$ molecules/cm$^3$.

By changing the time at which the image is taken, the motion of the cloud of molecules through the decelerator can be followed. Panel (b) of Figure 4 shows a series of 5 μs-wide images taken at 20 μs intervals, starting at 525.5 μs after the laser and with 5 μs wide gates. The time delay between subsequent images is 20 μs, corresponding to 1/2 period at 25 kHz. Above and below the images are plots of the electric field magnitude which has the same shape for each image. Vertical dashed lines show the calculated positions of the traps to guide the eye.

B. Guiding of molecules with static fields

Errors in the operation of the decelerator can stem from imperfections in the construction and in the waveforms. By applying static potentials to the electrodes, the errors due to imperfections in the construction can be examined independent of the waveform errors. If alternating positive and negative potentials are applied to the electrodes, the resulting distribution of electric field magnitude has a minimum on axis (shown in panel (a) of Figure 5) and increases steeply near the walls of the tube. This electric field distribution generates a guide that confines the molecules in low field seeking states transversely only.

Although there is also a longitudinal modulation of the electric field strength (including a point of zero electric field at the center of each electrode), this modulation is too weak to significantly affect the longitudinal motion of the molecules; application of guiding potentials increases the flux without significantly changing the arrival time distribution. The transverse position distribution several centimeters behind the decelerator, on the other hand, shows significant structure that can be used to characterize the decelerator. To measure this distribution, an imaging detector (microchannel plate (MCP) detector, coupled to a phosphor screen) is placed 18 cm behind the last electrode of the decelerator. Metastable CO molecules that hit the MCP generate an Auger electron which is subsequently amplified by the MCP stack. The resulting spot on the phosphor screen is recorded on a CCD camera. The center position of each spot is determined using a
centroiding algorithm. By switching the voltage applied to the front of the MCP stack, detection is restricted to molecules with an arrival time corresponding to a longitudinal velocity of $300 \pm 6 \text{ m/s}$.

The images in Figure 5(b) show the position distributions of molecules at the MCP with 0 kV, ±1 kV, ±2 kV, or ±3 kV applied to the electrodes of the decelerator. At zero volts (upper left image in Figure 5(b)), the molecules form a circular distribution, the size of which is mainly determined by the geometric aperture formed by the diameter and length of the guide, and the distance to the imaging detector. Two main changes are observed as the potential on the electrodes increases: (1) The overall intensity increases, and (2) in addition to the intense central disk, larger rings appear that grow with increasing voltage. The molecules execute oscillations in the strongly anharmonic potential, with an oscillation period that depends on the initial position in radial phase space. The packet enters the decelerator with a narrow transverse position distribution and a wide transverse velocity distribution. Because the faster molecules oscillate more rapidly than the slower molecules, the resulting transverse phase space distribution at the end of the decelerator is a spiral. The radial distribution in the image is a projection of the phase space distribution onto the transverse position coordinate, and a spiral-shaped distribution produces peaks with flatter sections in between. Increasing the voltage increases the transverse velocity acceptance, and transversely faster molecules will undergo more oscillations and produce larger rings. Each additional half-integer oscillation produces an additional ring in the images. While the images are almost circularly symmetric, some asymmetries are also visible. The prominent left-right asymmetry, particularly in the outer rings, results in part from an off-center placement of the laser excitation at the beginning of the experiment, and it can be influenced by moving the laser. An up-down asymmetry of a similar degree is obtained by slightly detuning the laser. This detuning leads to the selective excitation of molecules with a velocity component along the laser beam that are Doppler-shifted into resonance. The bandwidth of the laser corresponds to a velocity range of $\sim 30 \text{ m/s}$ which is in a range similar to the acceptance of the guide. A subtle eight-fold symmetry is also visible in some of the rings, which results from the eight different positions of the electrode indentations. These result from the shape that was chosen for the ring electrodes here (see Figure 2(d)). The subtlety of this structure in the images indicates that the indentations do not significantly affect the operation of the decelerator.

The left (right) panel of Figure 5(c) shows the experimental (calculated) angle-integrated radial intensity profile of the images in panel (b). The calculations were done via two-dimensional trajectory simulations that assume cylindrical symmetry. In all cases except with no voltage applied, the relative number of molecules reaching the detector is accurately reproduced by the trajectory simulations. In the field-free case, the lower number of molecules reaching the detector in the simulations can be qualitatively explained by the existence of non-low-field-seeking states; these states are present in the experiment and travel along the same straight-line trajectories as the molecules in the low field seeking state when no electric fields are applied, but are not included in the trajectory simulations. The ring diameters are accurately reproduced by the simulations, but the simulated rings have sharper edges than are observed experimentally. Since the peaks at the ring radii are wider in the experimental data than in the simulated images, the contrast is also reduced. As a consequence some of the rings, in particular the outermost ones, are barely visible in the experimental images while they are still clearly visible in the simulations. Simulations show that the rings stem from molecules that pass close to the electrodes at the end of the decelerator, and that are slightly focused. The intensity distribution in these rings depends sensitively on the electric field distribution just behind the decelerator. This region is more strongly influenced by the support structure of the electrodes, which is not included in the cylindrically-symmetric model.

Overall, the measurements show only minor deviations from the predictions of the two-dimensional trajectory simulations, indicating that the misalignments of the decelerator are relatively small. Remaining differences are attributed mostly to non-cylindrically symmetric fringe-fields behind the decelerator. These distort the transverse velocity distribution of the molecules emerging from the decelerator by asymmetric beam-focusing or -defocusing. They have not been included in the present calculations, and to do so would indeed be quite demanding, both because the fields would have to be accurately known, and because the calculations would have to be done in three dimensions, without assuming cylindrical symmetry.

C. Guiding of molecules with AC fields

Similar images to those shown in Sec. III B were also recorded after actively guiding the molecules at 288 m/s. Panel (a) of Figure 6 shows the angle-integrated radial
intensity profiles of the transverse distribution for the range of arrival times covering the fast part of an initial distribution centered around \(\sim 300 \text{ m/s}\). The main panel shows the distributions measured in 40 \(\mu\text{s}\)-wide arrival time gates over a large range. Such slices are obtained by applying a short voltage-pulse to the front of the MCP-stack in the imaging detector. The fast part of the initial molecular beam, which arrives around 2.2–2.3 ms, shows a narrow transverse distribution; these molecules move faster than the traps and are not confined by the longitudinal potential. They traverse the saddle points between traps and thus see a much weaker transverse confining potential. The inset of panel (a) shows the narrow range of arrival times below it with a higher resolution (3 \(\mu\text{s}\) wide gates). The narrow peaks that result as the molecules are bunched while traversing the saddle points are visible here. The longitudinally confined guided molecules are visible at an arrival time of 2.4 ms. Because these molecules are confined in a deep potential well, a much wider transverse velocity distribution can be guided to the end of the decelerator. Panel B shows the results of numerical trajectory simulations that agree well with the experimental data.

D. Non-adiabatic losses

In the Stark decelerator on a chip, it was observed that non-adiabatic transitions from the low field seeking states to states only weakly affected by the electric fields result in significant losses of molecules from the traps. Similar losses have been observed for ammonia molecules trapped in macroscopic electrostatic traps, albeit on a longer time scale.\(^{34}\) The chip decelerator operates on a similar principle as the one described here, in that the required fields for the deceleration are obtained via the application of sine-modulated potentials to a periodic electrode-array. It has been shown that, in order to decelerate molecules on this chip, the non-adiabatic losses must be prevented by either using the \(^{13}\text{CO}\) isotopologue\(^{26}\) or by applying an external magnetic field.\(^{35}\) Non-adiabatic losses of molecules from the traps on the chip result primarily from the small size of the traps; as the molecules pass near the zero-field trap center, the direction of the electric field vector changes rapidly, providing spectral components that can drive the transitions to non-trapped states. In the macroscopic traps of the decelerator presented here the rate of such transitions should be much lower.

To confirm that the effect of such losses is small, \(^{12}\text{C}^{16}\text{O}\) and \(^{13}\text{C}^{16}\text{O}\) were both guided at 300 m/s and decelerated from 300 m/s to 180 m/s over the length of the decelerator, and the arrival time distribution of the molecules exiting the decelerator was recorded. Because of the hyperfine structure in \(^{13}\text{CO}\), any losses due to non-adiabatic transitions should be reduced in comparison with \(^{12}\text{CO}\).\(^{26}\)

Figure 7(a) shows, from bottom to top, the experimental arrival time distribution for \(^{13}\text{CO}\) and \(^{12}\text{CO}\), and the simulation for \(^{12}\text{CO}\) when guiding of the molecules at 300 m/s. In all three traces, the main peak shows the molecules that are moving in the main trap, at 300 m/s, and the sharp and regular structures to either side of the main peak are from molecules that are bunched as they cross the saddle points between the traps. Except for an overall intensity difference, the three traces show no significant differences. The source of the different intensities lies mainly in different source conditions. Additional information on potential non-adiabatic losses can be obtained by comparing the traces for deceleration: this is shown in Fig. 7(b) for \(^{13}\text{CO}\) (lower trace) and \(^{12}\text{CO}\) (upper trace). By normalizing these traces based on the relative intensities of the guiding data, it is found that about 60% more \(^{12}\text{CO}\) is decelerated than \(^{13}\text{CO}\). This difference in deceleration efficiency is qualitatively explained by the slightly lower mass of \(^{12}\text{CO}\). Significant non-adiabatic losses would lead to an enhanced efficiency of \(^{13}\text{CO}\) deceleration.

Monte-Carlo trajectory simulations for guiding of \(^{12}\text{CO}\) (shown as uppermost trace of Figure 7(a)) and \(^{13}\text{CO}\) (not shown) show excellent agreement with the experimental data. The relative amounts of \(^{12}\text{CO}\) and \(^{13}\text{CO}\) that can be decelerated also agree well with the predictions of the simulations, which do not allow for non-adiabatic processes. On the timescale of the current experiments (which is similar to the radiative lifetime of the metastable molecular state that is used), losses due to non-adiabatic transitions are, thus, concluded to be negligible for the case of CO.

IV. SUMMARY AND OUTLOOK

A detailed account of the design, construction, and operation of the recently developed traveling wave decelerator for polar neutral molecules has been given. The very high accuracy required for the mechanical design calls for special care during the production and alignment of the individual electrodes. Generating high-voltage sinusoidal waveforms with frequencies between 25 kHz (for the current geometry) and ideally 0 Hz is particularly challenging. In the currently employed design, the range 12–26 kHz is accessible, and this could be extended by using different transformers. To decelerate molecules from higher initial velocities or to lower final velocities one could segment the decelerator into different parts that are each driven by a separate set of transformers. By
having slightly overlapping frequency ranges between neighboring segments, a smooth deceleration would then be possible over wider velocity ranges, albeit at the cost of flexibility. A more flexible solution, that, however, requires more development, will be the use of a linear amplifier that covers the entire range of interest.

The good agreement between the simulations and experiments raises the confidence in the understanding of the overall dynamics of molecules inside this type of decelerator. The longitudinal distribution is measured by recording the molecule arrival times, and the transverse distribution has been characterized via imaging of the complete packet of guided molecules in dc-electric fields. Narrow time-slices of the transverse distributions have been recorded for guiding molecules in dc-electric fields. Narrow time-slices of modulated fields, and it was found that the calculated and measured data agree reasonably well. Remaining discrepancies, for dc- and ac-guided molecules, are attributed to fringe fields that are generated from the electrode mounting structure. These fields are shielded from molecules that are inside the decelerator, but not from molecules flying from the end of the decelerator to the detector. The comparison between $^{12}\text{CO}$ and $^{13}\text{CO}$ showed that in the present geometry, unlike the case of the chip decelerator, non-adiabatic losses can be completely neglected. This is because the increased trap size leads to lower additional frequencies perceived by the molecules as they fly through the central region of the traps.

As has been suggested previously, this type decelerator seems particularly suitable for the deceleration of heavy diatomic molecules. Each rotational state of these molecules has low-field-seeking components only below a certain electric field. Above that, all levels undergo an avoided crossing with levels from the next rotational state and become high-field-seekers, making them inaccessible by traditional Stark deceleration. The only option so far was the use of alternating-gradient deceleration, but this approach has not yet been demonstrated to work stably and efficiently. In the present decelerator, however, the electric field magnitude inside the moving traps is always relatively low, yet the high electric field gradient nevertheless enables efficient deceleration.

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